

WILLIAMSON CREEK AQUIFER SUMMARY, 2008 **AQUIFER SAMPLING AND ASSESSMENT PROGRAM**



APPENDIX 11 TO THE 2009 TRIENNIAL SUMMARY REPORT
PARTIAL FUNDING PROVIDED BY THE CWA



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BACKGROUND

The Louisiana Department of Environmental Quality's (LDEQ) Aquifer Sampling and Assessment Program (ASSET) is an ambient monitoring program established to determine and monitor the quality of ground water produced from Louisiana's major freshwater aquifers. The ASSET Program samples approximately 200 water wells located in 14 aquifers and aquifer systems across the state. The sampling process is designed so that all fourteen aquifers and aquifer systems are monitored on a rotating basis, within a three-year period so that each well is monitored every three years.

In order to better assess the water quality of a particular aquifer, an attempt is made to sample all ASSET Program wells producing from it in a narrow time frame. To more conveniently and economically promulgate those data collected, a summary report on each aquifer is prepared separately. Collectively, these aquifer summaries will make up, in part, the ASSET Program's Triennial Summary Report for 2009.

Analytical and field data contained in this summary were collected from wells producing from the Williamson Creek aquifer during the 2008 state fiscal year (July 1, 2007 - June 30, 2009). This summary will become Appendix II of ASSET Program Triennial Summary Report for 2009.

These data show that in July 2008, seven wells were sampled which produce from the Williamson Creek aquifer. Three of these seven are classified as public supply, while two are classified as domestic and two are industrial. The wells are located in four parishes in central and southwest areas of the state.

Figure 11-1 shows the geographic locations of the Williamson Creek aquifer and the associated wells, whereas Table 11-1 lists the wells in the aquifer along with their total depths, use made of produced waters and date sampled.

Well data, including well location and aquifer assignment, for registered water wells were obtained from the Louisiana Department of Transportation and Development's Water Well Registration Data file.

GEOLOGY

The Williamson Creek member consists of sands, silts, silty clays, and some gravel. The Williamson Creek member, along with the Carnahan Bayou and Dough Hills, is grouped into the Jasper aquifer. The aquifer unit consists of fine to coarse sand, which may grade laterally and vertically to silt and clay.

HYDROGEOLOGY

Recharge takes place primarily as a result of the direct infiltration of rainfall in interstream, upland outcrop areas, movement of water through overlying terrace deposits, and leakage from other aquifers. The hydraulic conductivity of the Williamson Creek varies between 20-260 feet/day.

The maximum depths of occurrence of freshwater in the Williamson Creek range from 175 feet above sea level, to 2,450 feet below sea level. The range of thickness of the fresh water interval in the Williamson Creek is 50 to 1,250 feet. The depths of the Williamson Creek wells that were monitored in conjunction with the ASSET Program range from 190 to 1,657 feet.

PROGRAM PARAMETERS

The field parameters checked at each ASSET well sampling site and the list of conventional parameters analyzed in the laboratory are shown in Table 11-2. The inorganic (total metals) parameters analyzed in the laboratory are listed in Table 11-3. These tables also show the field and analytical results determined for each analyte. For quality control, duplicate samples were taken for each parameter at R-932.

In addition to the field, conventional and inorganic analytical parameters, the target analyte list includes three other categories of compounds: volatiles, semi-volatiles, and pesticides/PCBs. Due to the large number of analytes in these categories, tables were not prepared showing the analytical results for these compounds. A discussion of any detections from any of these three categories, if necessary, can be found in their respective sections. Tables 11-8, 11-9 and 11-10 list the target analytes for volatiles, semi-volatiles and pesticides/PCBs, respectively.

Tables 11-4 and 11-5 provide a statistical overview of field and conventional data, and inorganic data for the Williamson Creek aquifer, listing the minimum, maximum, and average results for these parameters collected in the FY 2008 sampling. Tables 11-6 and 11-7 compare these same parameter averages to historical ASSET-derived data for the Williamson Creek aquifer, from fiscal years 1996, 2000, 2003 and 2006.

The average values listed in the above referenced tables are determined using all valid, reported results, including non-detects. Per Departmental policy concerning statistical analysis, one-half of the detection limit (DL) is used in place of zero when non-detects are encountered. However, the minimum value is reported as less than the DL, not one-half the DL. If all values for a particular analyte are reported as non-detect, then the minimum, maximum, and average values are all reported as less than the DL. For contouring purposes, one-half the DL is also used for non-detects in the figures and charts referenced below.

Figures 11-2, 11-3, 11-4, and 11-5, respectively, represent the contoured data for pH, total dissolved solids (TDS), chloride (Cl) and iron. Charts 11-1 through 11-16 represent the trend of the graphed parameter, based on the averaged value of that parameter for each three-year reporting period. Discussion of historical data and related trends is found in the **Water Quality Trends and Comparison to Historical ASSET Data** section.

INTERPRETATION OF DATA

Under the Federal Safe Drinking Water Act, EPA has established maximum contaminant levels (MCLs) for pollutants that may pose a health risk in public drinking water. An MCL is the highest level of a contaminant that EPA allows in public drinking water. MCLs ensure that drinking water does not pose either a short-term or long-term health risk. While not all wells sampled were public supply wells, the Office of Environmental Assessment does use the MCLs as a benchmark for further evaluation.

EPA has set secondary standards, which are defined as non-enforceable taste, odor, or appearance guidelines. Field and laboratory data contained in Tables 11-2 and 11-3 show that only two secondary MCL (SMCL) was exceeded in 2 of the 7 wells sampled in the Williamson Creek aquifer.

Field and Conventional Parameters

Table 11-2 shows the field and conventional parameters for which samples are collected at each well and the analytical results for those parameters. Table 11-4 provides an overview of this data for the Williamson Creek aquifer, listing the minimum, maximum, and average results for these parameters.

Federal Primary Drinking Water Standards: A review of the analysis listed in Table 11-2 shows that no primary MCL was exceeded for field or conventional parameters for this reporting period. The ASSET well reporting turbidity level greater than 1.0 NTU does not exceed the Primary MCL of 1.0, as this standard applies to public supply water wells that are under the direct influence of surface water. The Louisiana Department of Health and Hospitals has determined that no public water supply well in Louisiana was in this category.

Federal Secondary Drinking Water Standards: A review of the analysis listed in Table 11-2 shows that one well exceeded the SMCL for pH. Laboratory results override field results in exceedance determination, thus only laboratory results will be counted in determining SMCL exceedance numbers. Following is a list of SMCL parameter exceedances with well number and results:

pH (SMCL = 6.5 – 8.5 Standard Units):

BE-407 – 8.51 SU

Inorganic Parameters

Table 11-3 shows the inorganic (total metals) parameters for which samples are collected at each well and the analytical results for those parameters. Table 11-5 provides an overview of inorganic data for the Williamson Creek aquifer, listing the minimum, maximum, and average results for these parameters.

Federal Primary Drinking Water Standards: A review of the analyses listed on Table 11-3 shows that no primary MCL was exceeded for total metals.

Federal Secondary Drinking Water Standards: Laboratory data contained in Table 11-3 shows that 1 well exceeded the secondary MCL for iron:

Iron (SMCL = 300 ug/L):

CO-163 – 677 ug/L

Volatile Organic Compounds

Table 11-8 shows the volatile organic compound (VOC) parameters for which samples are collected at each well. Due to the number of analytes in this category, analytical results are not tabulated; however, any detection of a VOC would be discussed in this section.

No VOCs were detected at or above their respective detection limits during the FY 2008 sampling of the Williamson Creek aquifer.

Semi-Volatile Organic Compounds

Table 11-9 shows the semi-volatile organic compound (SVOC) parameters for which samples are collected at each well. Due to the number of analytes in this category, analytical results are not tabulated; however any detection of a SVOC would be discussed in this section.

There were no confirmed SVOC detections at or above its detection limit during the FY 2008 sampling of the Williamson Creek aquifer.

Pesticides and PCBs

Table 11-10 shows the pesticide and PCB parameters for which samples are collected at each well. Due to the number of analytes in this category, analytical results are not tabulated; however any detection of a pesticide or PCB would be discussed in this section.

No pesticide or PCB was detected at or above its detection limit during the FY 2008 sampling of the Williamson Creek aquifer.

WATER QUALITY TRENDS AND COMPARISON TO HISTORICAL ASSET DATA

Analytical and field data show that the quality and characteristics of ground water produced from the Williamson Creek aquifer exhibit some changes when comparing current data to that of the four previous sampling rotations (three, six, nine and twelve years prior). These comparisons can be found in Tables 11-6 and 11-7, and in Charts 11-1 to 11-16 of this summary. Over the twelve-year period, 7 analytes have shown a general increase in average concentration. These analytes are: temperature, specific conductance (field and lab), total dissolved solids (TDS), TKN, hardness, sulfate, and alkalinity. For this same time period, 5 analytes have demonstrated a decrease in average concentration: color, nitrite-nitrate, zinc, total phosphorus (P), and pH. Iron, ammonia, chloride, salinity and copper have remained consistent for this time period. Barium initially showed a decrease in concentration but has remained consistent since 2003.

The number of secondary exceedances in the Williamson Creek aquifer continued to remain low. The previous sampling FY 2006 showed three SMCL exceedances, while there is only one SMCL exceedance in FY 2008 sampling.

SUMMARY AND RECOMMENDATIONS

In summary, the data show that the ground water produced from this aquifer is soft¹ and is of good quality when considering short-term or long-term health risk guidelines. Laboratory data show that no ASSET well that was sampled during the Fiscal Year 2008 monitoring of the Williamson Creek aquifer exceeded a Primary MCL. The data also show that this aquifer is of good quality when considering taste, odor, or appearance guidelines, with one Secondary MCLs exceeded in one well.

Comparison to historical ASSET-derived data shows some change in the quality or characteristics of the Williamson Creek aquifer, with 7 parameters showing consistent increases in concentration, 5 parameters decreasing in concentration, and 5 parameters showing no consistent change over the previous twelve years.

It is recommended that the wells assigned to the Williamson Creek aquifer be re-sampled as planned, in approximately three years. In addition, several wells should be added to the seven currently in place to increase the well density for this aquifer.

¹ Classification based on hardness scale from: Peavy, H. S. et al. *Environmental Engineering*. New York: McGraw-Hill, 1985.

Table 11-1: List of Wells Sampled, Williamson Creek Aquifer–FY 2008

DOTD Well Number	Parish	Date	Owner	Depth (Feet)	Well Use
BE-407	BEAUREGARD	7/22/2008	BOISE CASCADE	1,657	INDUSTRIAL
CO-163	CONCORDIA	7/21/2008	U. S. ARMY CORPS OF ENG.	513	PUBLIC SUPPLY
R-867	RAPIDES	7/21/2008	INTERNATIONAL PAPER CO.	385	INDUSTRIAL
R-932	RAPIDES	7/21/2008	CITY OF ALEXANDRIA	466	PUBLIC SUPPLY
V-420	VERNON	7/22/2008	U.S. ARMY/FORT POLK	920	PUBLIC SUPPLY
V-5858Z	VERNON	7/21/2008	PRIVATE OWNER	248	DOMESTIC
V-8681Z	VERNON	7/22/2008	PRIVATE OWNER	190	DOMESTIC

Table 11-2: Summary of Field and Conventional Data, Williamson Creek Aquifer–FY 2008

DOTD Well Number	Temp Deg. C	pH SU	Sp. Cond. mmhos/cm	Sal. ppt	TDS g/L	Alk mg/L	Cl mg/L	Color PCU	Sp. Cond. umhos/cm	SO4 mg/L	TDS mg/L	TSS mg/L	Turb. NTU	NH3 mg/L	Hard. mg/L	Nitrite-Nitrate (as N) mg/L	TKN mg/L	Tot. P mg/L
	LABORATORY DETECTION LIMITS →					2.0	1.3	5	10	1.25	4	4	1	0.1	5.0	0.05	0.10	0.05
	FIELD PARAMETERS					LABORATORY PARAMETERS												
BE-407	31.30	8.51	0.394	0.19	0.26	208	8	<5	416	10.3	263	<4	<1	0.36	10.4	<0.05	0.41	0.17
CO-163	22.27	7.55	0.568	0.28	0.37	160	99.7	6	604	<1.25	352	<4	†1.9	0.48	28.7	<0.05	‡0.5	0.18
R-867	21.70	6.84	0.44	0.21	0.29	108	70.5	<5	476	26.2	300	<4	<1	0.3	34.3	<0.05	0.33	<0.05
R-932	23.08	8.19	0.427	0.20	0.28	237	12.5	<5	464	<1.25	276	<4	<1	0.3	18.1	<0.05	‡0.31	0.08
R-932*	23.08	8.19	0.427	0.20	0.28	237	12.9	<5	461	<1.25	276	<4	<1	0.27	17.9	<0.05	‡0.28	0.08
V-420	25.64	6.92	0.234	0.11	0.15	95.9	20.3	<5	250	5.6	208	<4	<1	0.35	18.1	<0.05	0.37	0.17
V-5858Z	24.61	8.01	0.444	0.21	0.29	159	58	<5	475	3.6	262	<4	<1	0.17	157	0.14	0.17	<0.05
V-8681Z	21.87	7.23	0.137	0.06	0.09	59.7	5.8	<5	142	5.4	143	<4	<1	0.21	23.9	<0.05	<0.1	0.44

*Denotes Duplicate Sample

†Estimated Value

‡Reported from a Dilution

Shaded cells exceed EPA Secondary Standards

Table 11-3: Summary of Inorganic Data, Williamson Creek Aquifer–FY 2008

DOTD Well Number	Antimony ug/L	Arsenic ug/L	Barium ug/L	Beryllium ug/L	Cadmium ug/L	Chromium ug/L	Copper ug/L	Iron ug/L	Lead ug/L	Mercury ug/L	Nickel ug/L	Selenium ug/L	Silver ug/L	Thallium ug/L	Zinc ug/L
Laboratory Detection Limits	1	3	2	1	0.5	3	3	20	3	0.05	3	4	0.5	1	10
BE-407	<1	<3	34.7	<1	<0.5	<3	<3	<20	<3	<0.05	<3	<4	<0.5	<1	<10
CO-163	<1	<3	94.2	<1	<0.5	<3	<3	677	<3	<0.05	<3	<4	<0.5	<1	466
R-867	<1	<3	75.5	<1	<0.5	<3	3.2	282	<3	<0.05	<3	<4	<0.5	<1	<10
R-932	<1	<3	49.1	<1	<0.5	<3	<3	<20	<3	<0.05	<3	<4	<0.5	<1	<10
R-932*	<1	<3	49.3	<1	<0.5	<3	<3	<20	<3	<0.05	<3	<4	<0.5	<1	<10
V-420	<1	<3	51.1	<1	<0.5	<3	<3	114	<3	<0.05	<3	<4	<0.5	<1	13.4
V-5858Z	<1	<3	326	<1	<0.5	<3	<3	167	<3	<0.05	<3	<4	<0.5	<1	<10
V-8681Z	<1	<3	41.6	<1	<0.5	<3	<3	29.8	<3	<0.05	<3	<4	<0.5	<1	<10

*Denotes Duplicate Sample

Shaded cells exceed EPA Secondary Standards

Table 11-4: FY 2008 Field and Conventional Statistics, ASSET Wells

	PARAMETER	MINIMUM	MAXIMUM	AVERAGE
FIELD	Temperature (°C)	21.7	31.3	24.19
	pH (SU)	6.84	8.51	7.68
	Specific Conductance (mmhos/cm)	0.137	0.568	0.38
	Salinity (ppt)	0.06	0.28	0.18
	TDS (g/L)	0.089	0.37	0.25
LABORATORY	Alkalinity (mg/L)	59.7	237	158.1
	Chloride (mg/L)	5.8	99.7	36
	Color (PCU)	<5	6	<5
	Specific Conductance (umhos/cm)	142	604	411
	Sulfate (mg/L)	<1.25	26.2	6.62
	TDS (mg/L)	143	352	260
	TSS (mg/L)	<4	<4	<4
	Turbidity (NTU)	<1	1.9	<1
	Ammonia, as N (mg/L)	0.17	0.48	0.31
	Hardness (mg/L)	10.4	157	38.6
	Nitrite - Nitrate, as N (mg/L)	<0.05	0.14	<0.05
	TKN (mg/L)	<0.1	0.5	0.3
	Total Phosphorus (mg/L)	<0.05	0.44	0.15

Table 11-5: FY 2008 Inorganic Statistics, ASSET Wells

PARAMETER	MINIMUM	MAXIMUM	AVERAGE
Antimony (ug/L)	<1	<1	<1
Arsenic (ug/L)	<3	<3	<3
Barium (ug/L)	34.7	326	90.2
Beryllium (ug/L)	<1	<1	<1
Cadmium (ug/L)	<0.5	<0.5	<0.5
Chromium (ug/L)	<3	<3	<3
Copper (ug/L)	<3	3.2	<3
Iron (ug/L)	<20	677	162.5
Lead (ug/L)	<3	<3	<3
Mercury (ug/L)	<0.05	<0.05	<0.05
Nickel (ug/L)	<3	<3	<3
Selenium (ug/L)	<4	<4	<4
Silver (ug/L)	<0.5	<0.5	<0.5
Thallium (ug/L)	<1	<1	<1
Zinc (ug/L)	<10	466	63.7

Table 11-6: Triennial Field and Conventional Statistics, ASSET Wells

PARAMETER		FY 1996 AVERAGE	FY 2000 AVERAGE	FY 2003 AVERAGE	FY 2006 AVERAGE	FY 2008 AVERAGE
FIELD	Temperature (°C)	23.82	23.12	24.00	25.27	24.19
	pH (SU)	6.86	7.83	7.54	Not Available	7.68
	Specific Conductance (mmhos/cm)	0.369	0.424	0.384	0.44	0.38
	Salinity (Sal.) (ppt)	0.18	0.20	0.18	0.21	0.18
	TDS (Total dissolved solids) (g/L)	-	-	-	-	0.25
LABORATORY	Alkalinity (Alk.) (mg/L)	136.1	150.3	139.6	153.8	158.1
	Chloride (Cl) (mg/L)	38.7	37.0	32.3	41.48	36
	Color (PCU)	12.1	5.0	<5	14.72	<5
	Specific Conductance (umhos/cm)	385.7	398.8	369.4	440.9	411
	Sulfate (SO4) (mg/L)	7.15	4.61	4.61	8.02	6.62
	TDS (Total dissolved solids) (mg/L)	211.3	272.7	235.7	284.8	260
	TSS (Total suspended solids) (mg/L)	<4	<4	<4	<4	<4
	Turbidity (Turb.) (NTU)	1.25	6.03	1.23	2.6	<1
	Ammonia, as N (NH3) (mg/L)	0.36	0.19	0.25	0.33	0.31
	Hardness (mg/L)	30.8	39.5	34.9	34.5	38.6
	Nitrite - Nitrate, as N (mg/L)	<0.05	0.15	<0.05	<0.05	<0.05
	TKN (mg/L)	0.32	0.40	0.39	0.70	0.3
	Total Phosphorus (P) (mg/L)	0.30	0.20	0.18	0.15	0.15

Table 11-7: Triennial Inorganic Statistics, ASSET Wells

PARAMETER	FY 1996 AVERAGE	FY 2000 AVERAGE	FY 2003 AVERAGE	FY 2006 AVERAGE	FY 2008 AVERAGE
Antimony (ug/L)	<5	<5	<5	<10	<1
Arsenic (ug/L)	<5	<5	<5	<10	<3
Barium (ug/L)	48.21	112.50	89.57	91.98	90.2
Beryllium (ug/L)	<1	<1	<1	<1	<1
Cadmium (ug/L)	<1	<1	<1	<1	<0.5
Chromium (ug/L)	<5	<5	<5	<5	<3
Copper (ug/L)	9.70	<5	<5	<10	<3
Iron (ug/L)	466.00	115.28	380.1	641.6	162.5
Lead (ug/L)	<10	<10	<10	<10	<3
Mercury (ug/L)	<0.05	<0.05	<0.05	<0.05	<0.05
Nickel (ug/L)	9.25	<5	<5	<5	<3
Selenium (ug/L)	<5	<5	<5	<5	<4
Silver (ug/L)	<1	<1	<1	<10	<0.5
Thallium (ug/L)	<5	<5	<5	<5	<1
Zinc (ug/L)	298.00	245.22	107.22	114.1	63.7

Table 11-8: VOC Analytical Parameters

COMPOUND	METHOD	DETECTION LIMIT (ug/L)
1,1-Dichloroethane	624	2
1,1-Dichloroethene	624	2
1,1,1-Trichloroethane	624	2
1,1,2-Trichloroethane	624	2
1,1,2,2-Tetrachloroethane	624	2
1,2-Dichlorobenzene	624	2
1,2-Dichloroethane	624	2
1,2-Dichloropropane	624	2
1,3- Dichlorobenzene	624	2
1,4-Dichlorobenzene	624	2
Benzene	624	2
Bromoform	624	2
Carbon tetrachloride	624	2
Chlorobenzene	624	2
Dibromochloromethane	624	2
Chloroethane	624	2
trans-1,2-Dichloroethene	624	2
cis-1,3-Dichloropropene	624	2
Bromodichloromethane	624	2
Methylene chloride	624	2
Ethyl benzene	624	2
Bromomethane	624	2
Chloromethane	624	2
o-Xylene	624	2
Styrene	624	2
Methylt-butyl ether	624	2
Tetrachloroethene	624	2
Toluene	624	2
trans-1,3-Dichloropropene	624	2
Trichloroethene	624	2
Trichlorofluoromethane	624	2
Chloroform	624	2
Vinyl chloride	624	2
Xylenes, m & p	624	4

Table 11-9: SVOC Analytical Parameters

COMPOUND	METHOD	DETECTION LIMIT (ug/L)
1,2-Dichlorobenzene	625	10
1,2,3-Trichlorobenzene	625	10
1,2,3,4-Tetrachlorobenzene	625	10
1,2,4-Trichlorobenzene	625	10
1,2,4,5-Tetrachlorobenzene	625	10
1,3-Dichlorobenzene	625	10
1,3,5-Trichlorobenzene	625	10
1,4-Dichlorobenzene	625	10
2-Chloronaphthalene	625	10
2-Chlorophenol	625	20
2-Methyl-4,6-dinitrophenol	625	20
2-Nitrophenol	625	20
2,4-Dichlorophenol	625	20
2,4-Dimethylphenol	625	20
2,4-Dinitrophenol	625	20
2,4-Dinitrotoluene	625	10
2,4,6-Trichlorophenol	625	20
2,6-Dinitrotoluene	625	10
3,3'-Dichlorobenzidine	625	10
4-Bromophenyl phenyl ether	625	10
4-Chloro-3-methylphenol	625	20
4-Chlorophenyl phenyl ether	625	10
4-Nitrophenol	625	20
Acenaphthene	625	10
Acenaphthylene	625	10
Anthracene	625	10
Benzidine	625	20
Benzo[a]pyrene	625	10
Benzo[k]fluoranthene	625	10
Benzo[a]anthracene	625	10
Benzo[b]fluoranthene	625	10
Benzo[g,h,i]perylene	625	10
Bis(2-chloroethoxy)methane	625	10
Bis(2-ethylhexyl)phthalate	625	10
Bis(2-chloroethyl)ether	625	10
Bis(2-chloroisopropyl)ether	625	10

Table 11-9: SVOCs (Continued)

COMPOUND	METHOD	DETECTION LIMIT (ug/L)
Butylbenzylphthalate	625	10
Chrysene	625	10
Dibenzo[a,h]anthracene	625	10
Diethylphthalate	625	10
Dimethylphthalate	625	10
Di-n-butylphthalate	625	10
Di-n-octylphthalate	625	10
Fluoranthene	625	10
Fluorene	625	10
Hexachlorobenzene	625	10
Hexachlorobutadiene	625	10
Hexachlorocyclopentadiene	625	10
Hexachloroethane	625	10
Indeno[1,2,3-cd]pyrene	625	10
Isophorone	625	10
Naphthalene	625	10
Nitrobenzene	625	10
N-Nitrosodimethylamine	625	10
N-Nitrosodiphenylamine	625	10
N-nitroso-di-n-propylamine	625	10
Pentachlorobenzene	625	10
Pentachlorophenol	625	20
Phenanthrene	625	10
Phenol	625	20
Pyrene	625	10

Table 11-10: Pesticides and PCBs

COMPOUND	METHOD	DETECTION LIMITS (ug/L)
4,4'-DDD	608	0.05
4,4'-DDE	608	0.05
4,4'-DDT	608	0.05
Aldrin	608	0.05
Alpha-Chlordane	608	0.05
alpha-BHC	608	0.05
beta-BHC	608	0.05
delta-BHC	608	0.05
gamma-BHC	608	0.05
Chlordane	608	0.2
Dieldrin	608	0.05
Endosulfan I	608	0.05
Endosulfan II	608	0.05
Endosulfan Sulfate	608	0.05
Endrin	608	0.05
Endrin Aldehyde	608	0.05
Endrin Ketone	608	0.05
Heptachlor	608	0.05
Heptachlor Epoxide	608	0.05
Methoxychlor	608	0.05
Toxaphene	608	2
Gamma-Chlordane	608	0.05
PCB-1016	608	1
PCB-1221	608	1
PCB-1232	608	1
PCB-1242	608	1
PCB-1248	608	1
PCB-1254	608	1
PCB-1260	608	1

Figure 11-1: Location Plat, Williamson Creek Aquifer

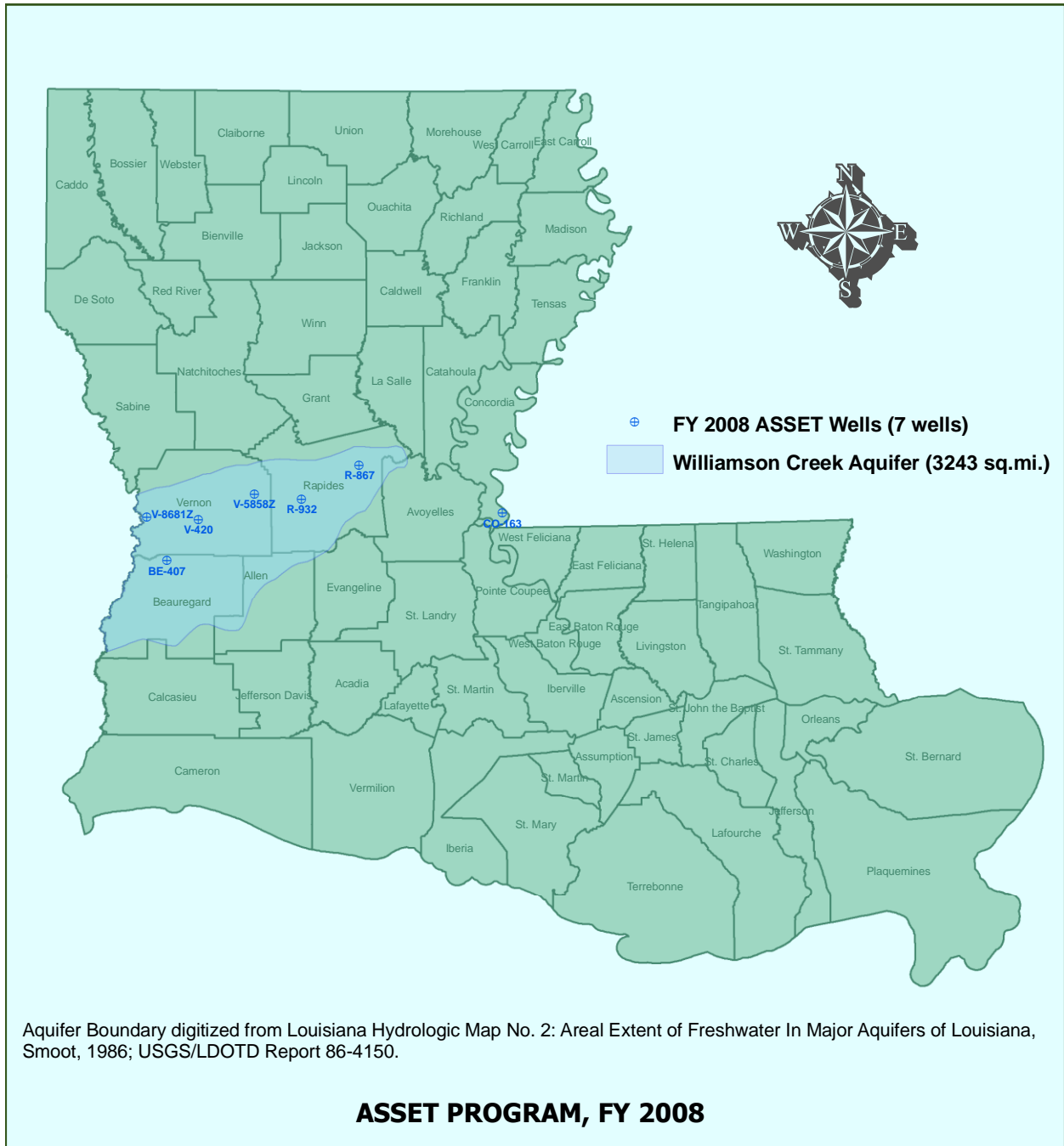


Figure 11-2: Map of pH Data

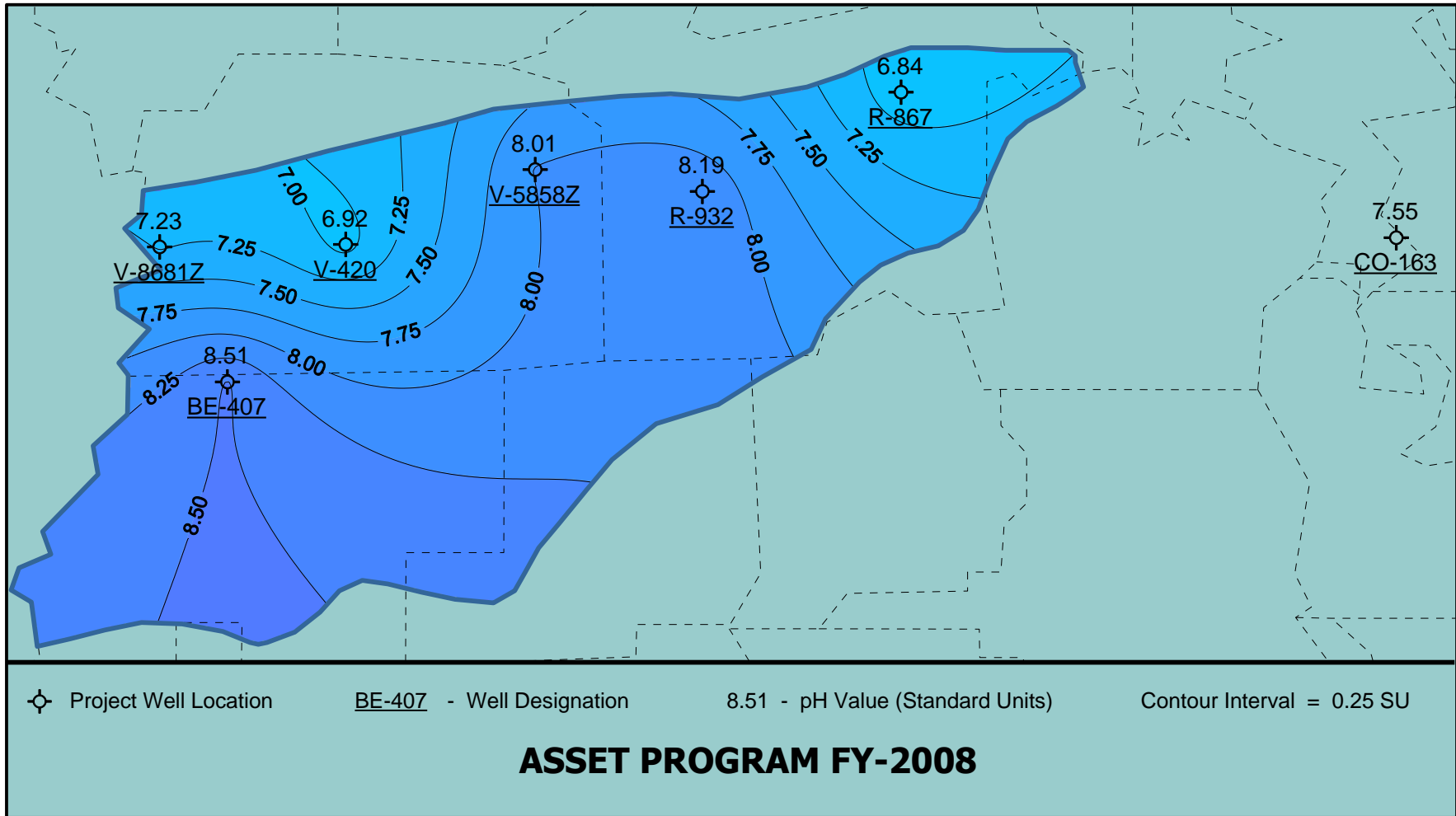


Figure 11-3: Map of TDS Lab Data

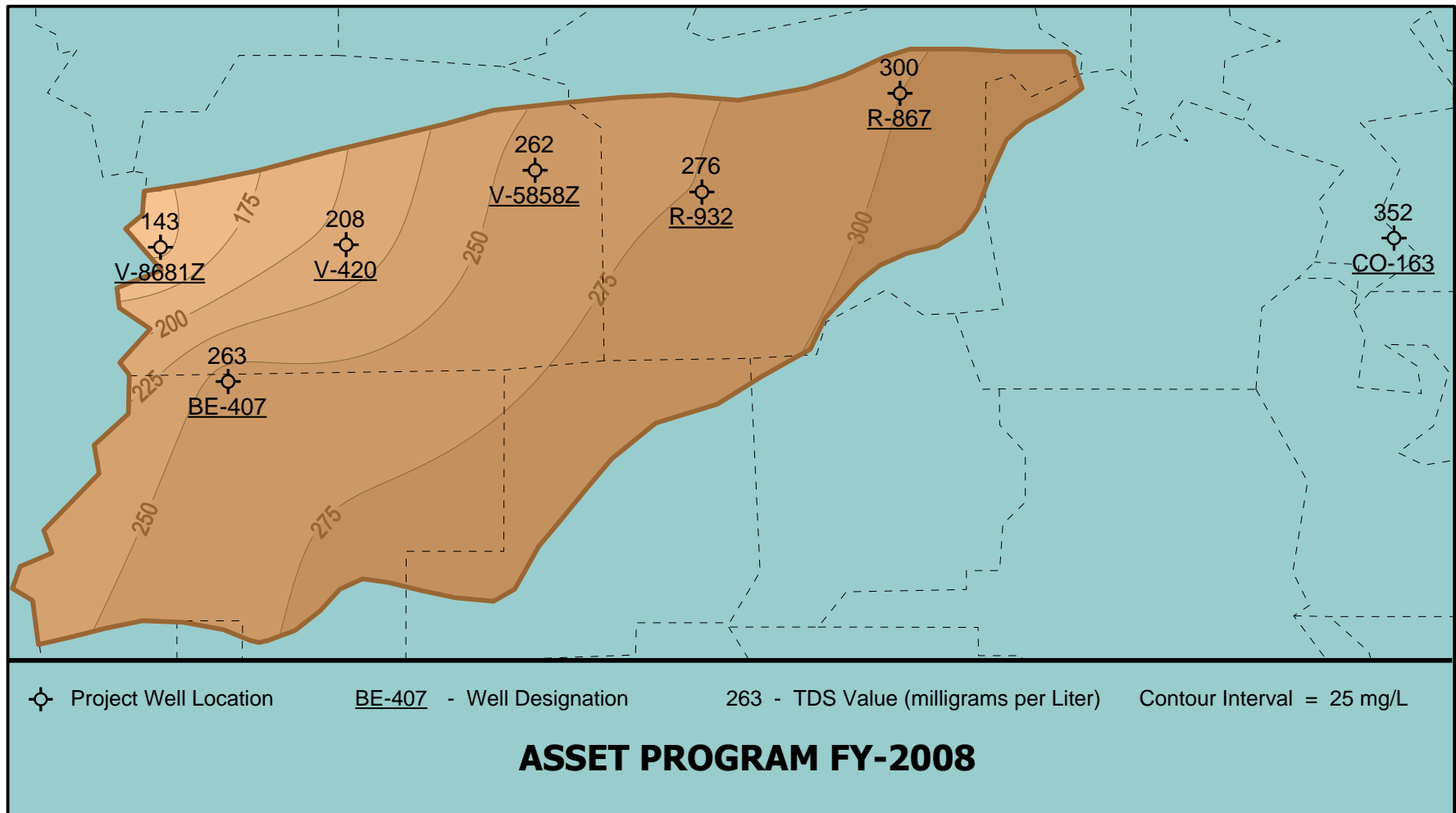


Figure 11-4: Map of Chloride Lab Data

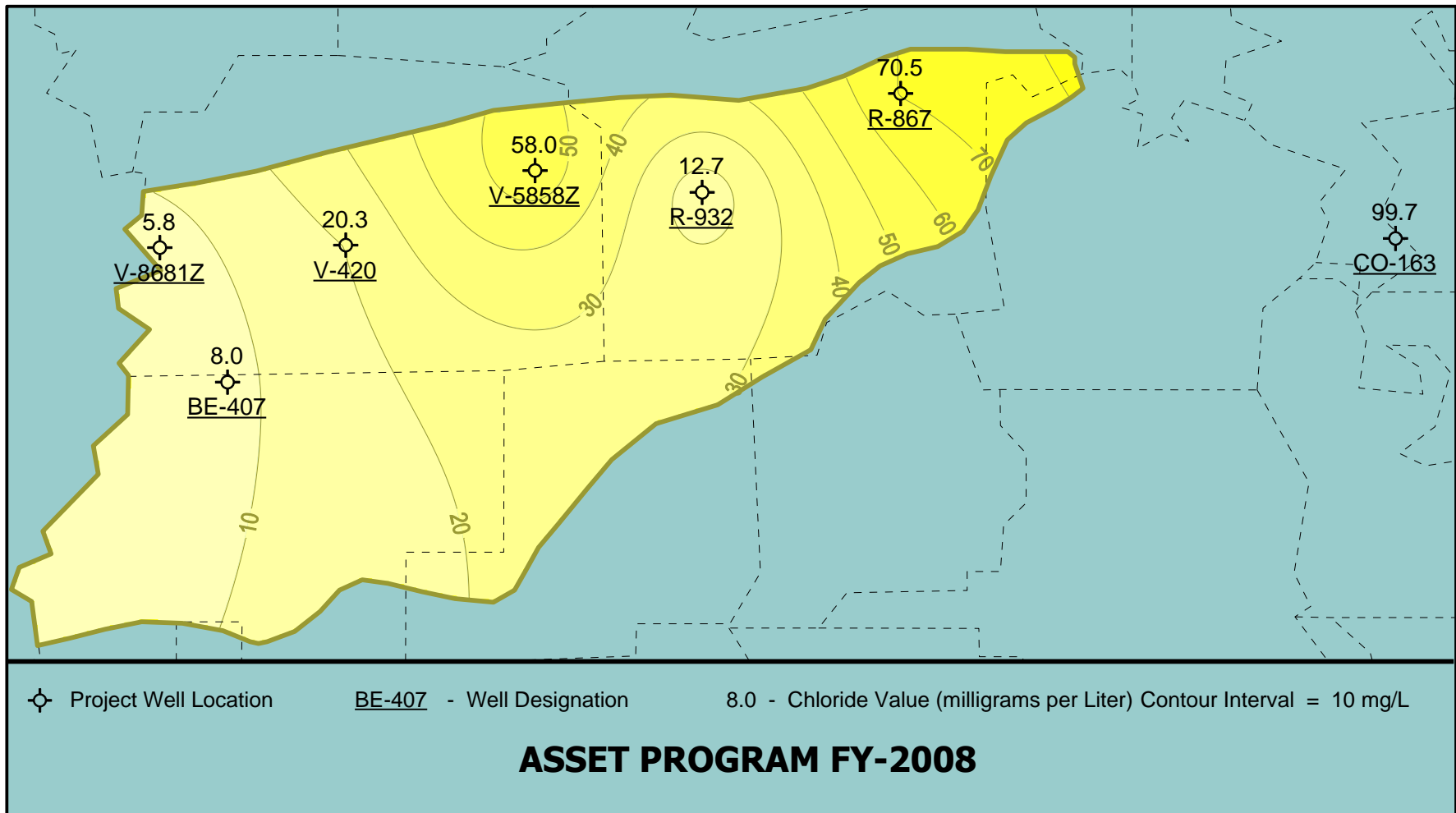


Figure 11-5: Map of Iron Data

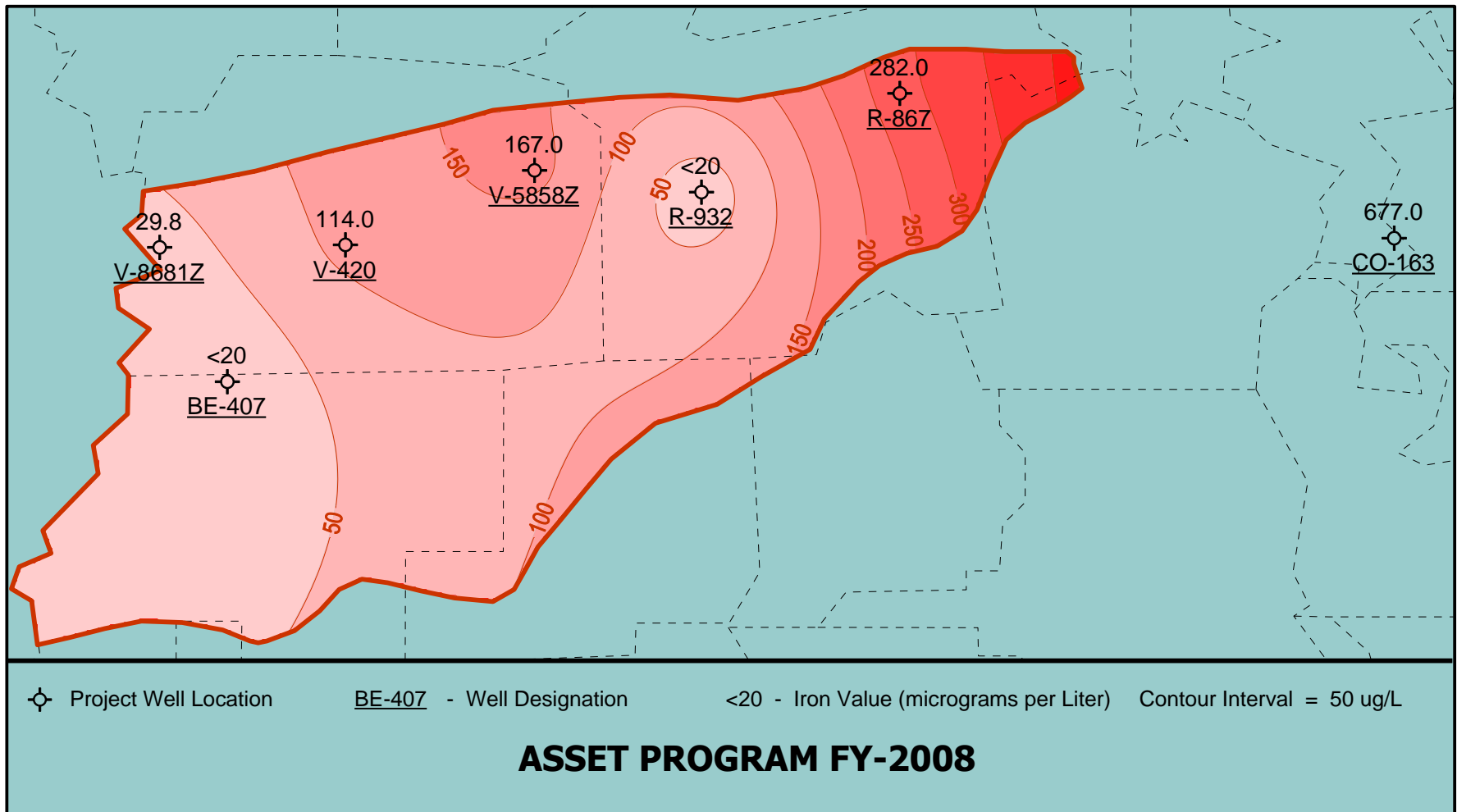


Chart 11-1: Temperature Trend

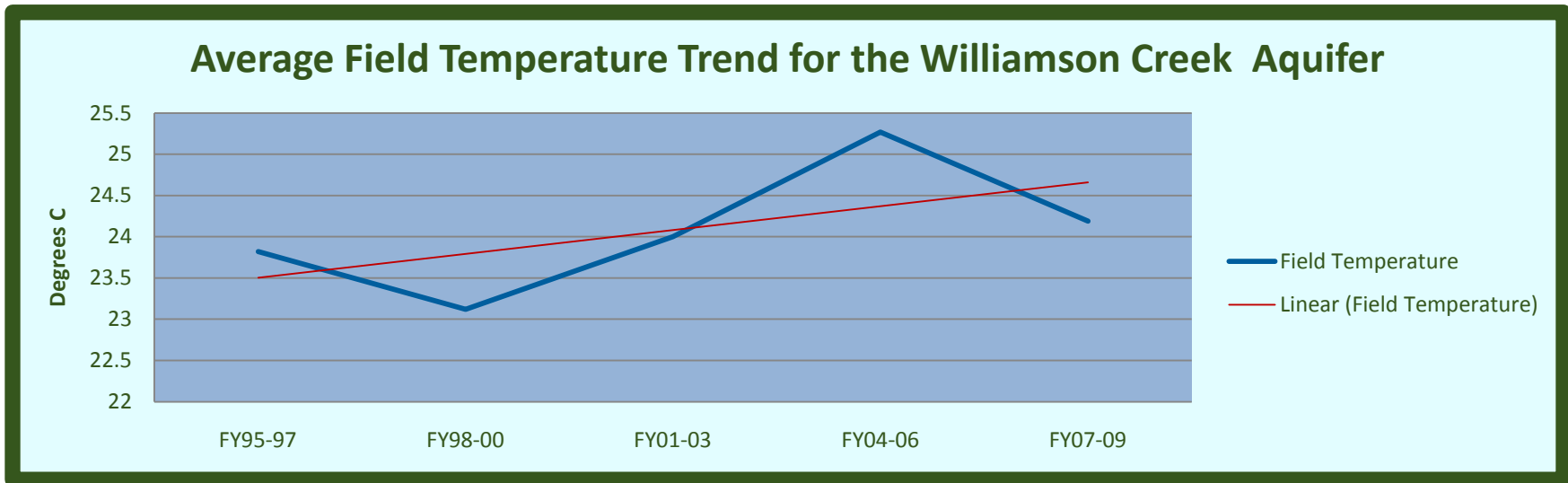


Chart 11-2: pH Trend

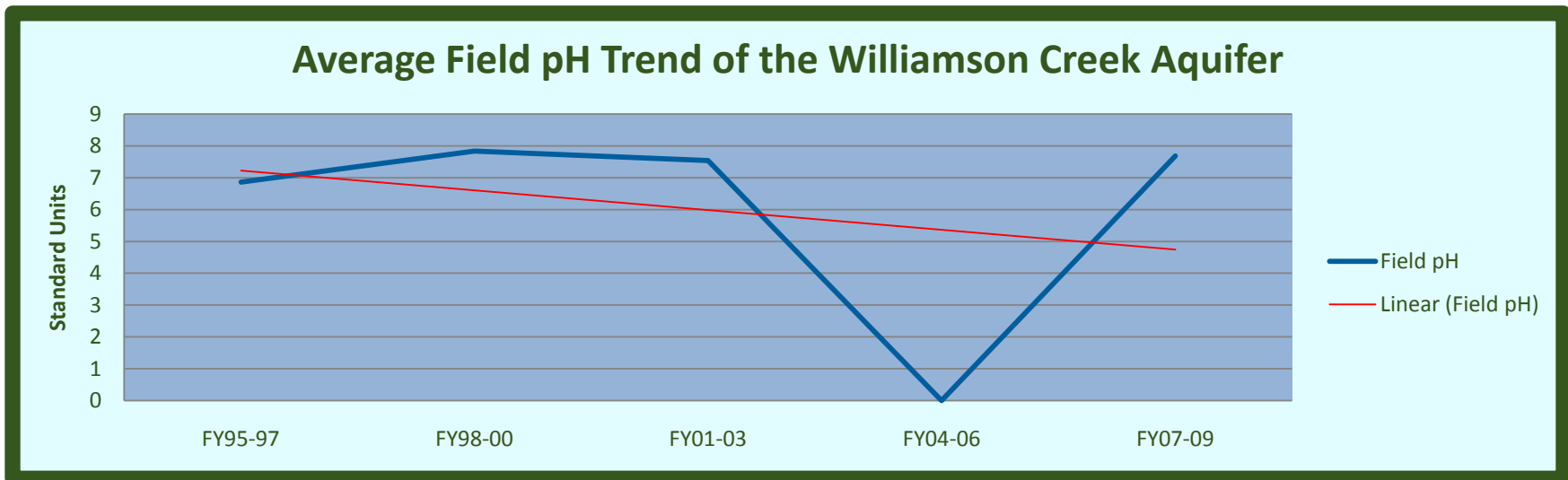


Chart 11-3: Field Specific Conductance Trend

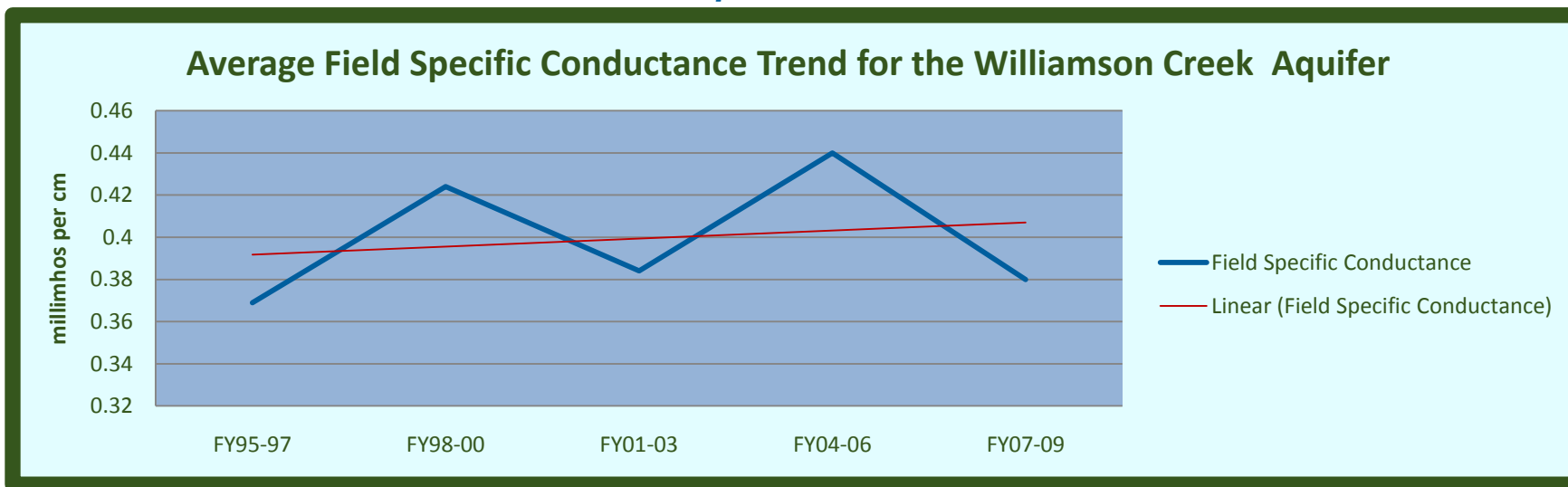


Chart 11-4: Lab Specific Conductance Trend

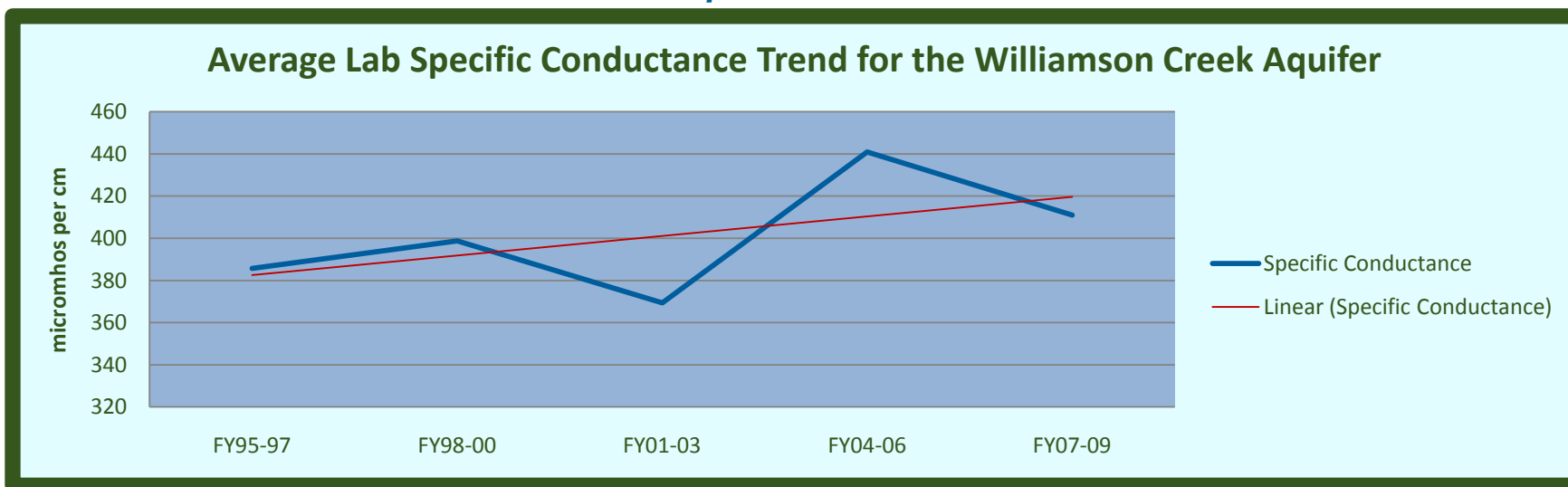


Chart 11-5: Field Salinity Trend

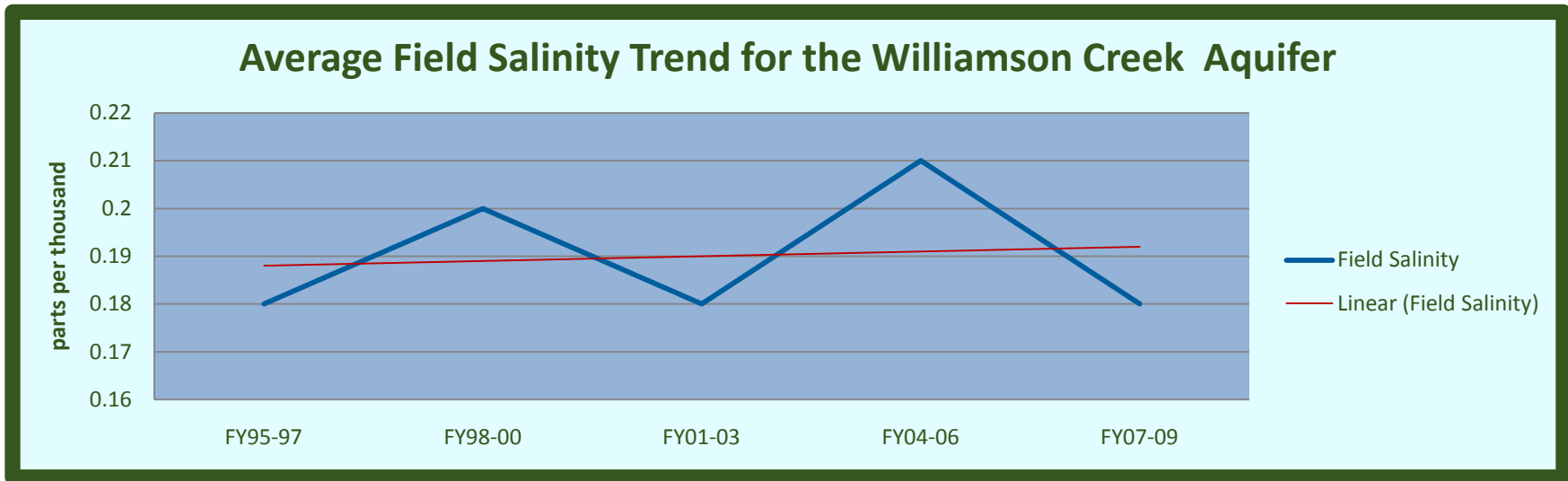


Chart 11-6: Alkalinity Trend

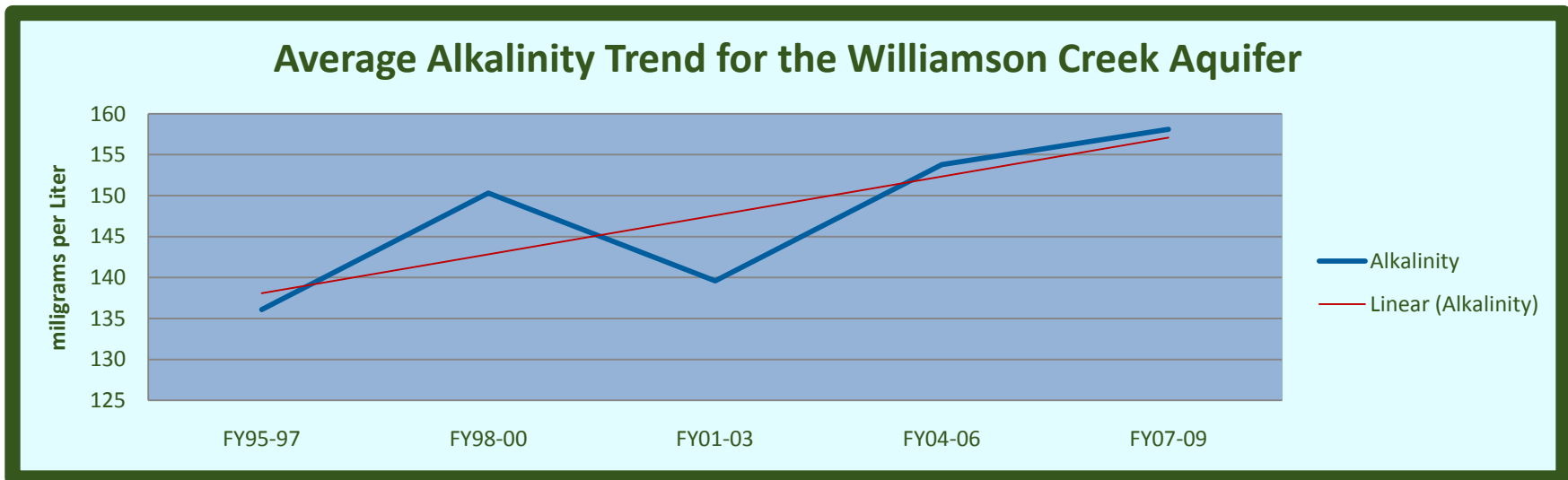


Chart 11-7: Chloride Trend

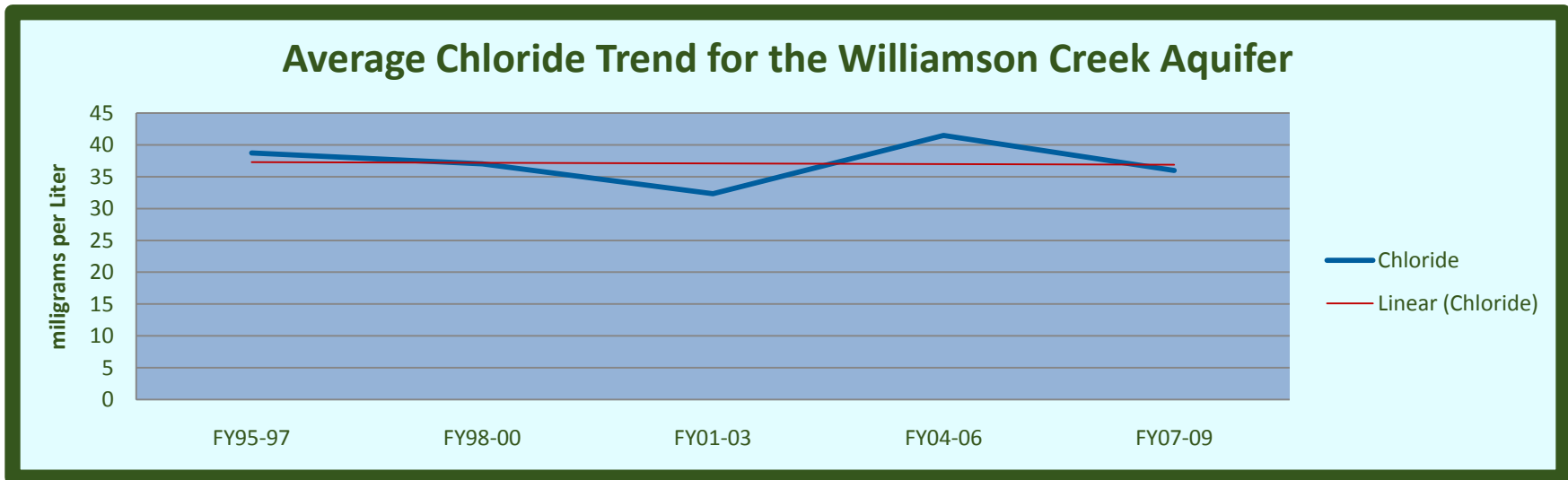


Chart 11-8: Color Trend

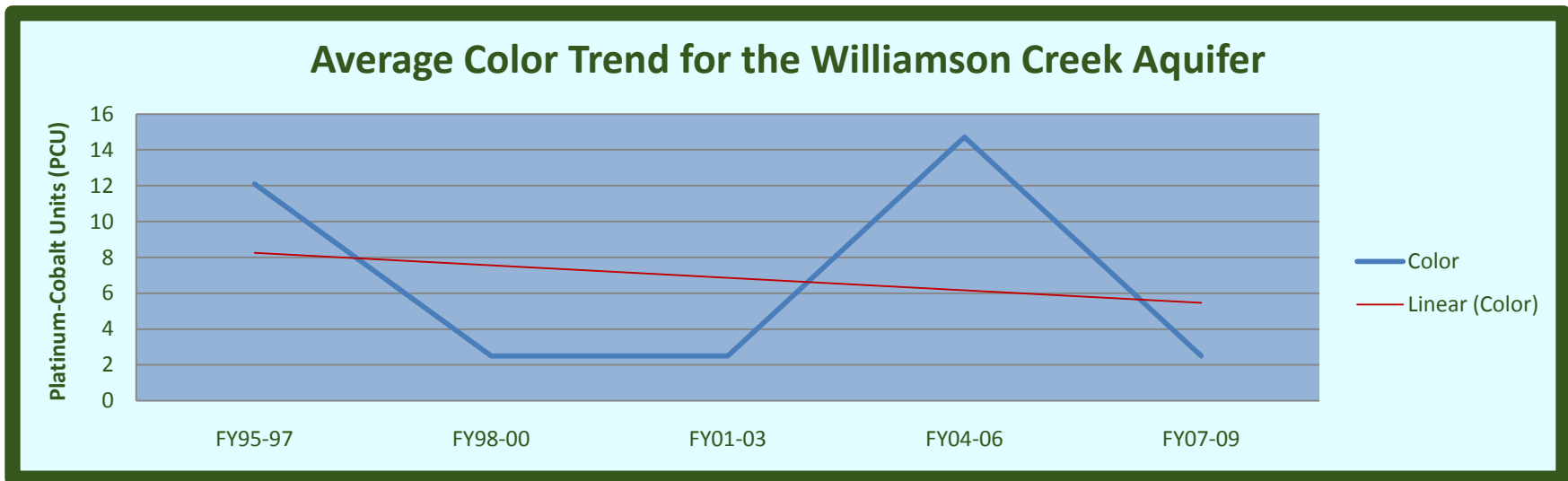


Chart 11-9: Sulfate (SO4) Trend

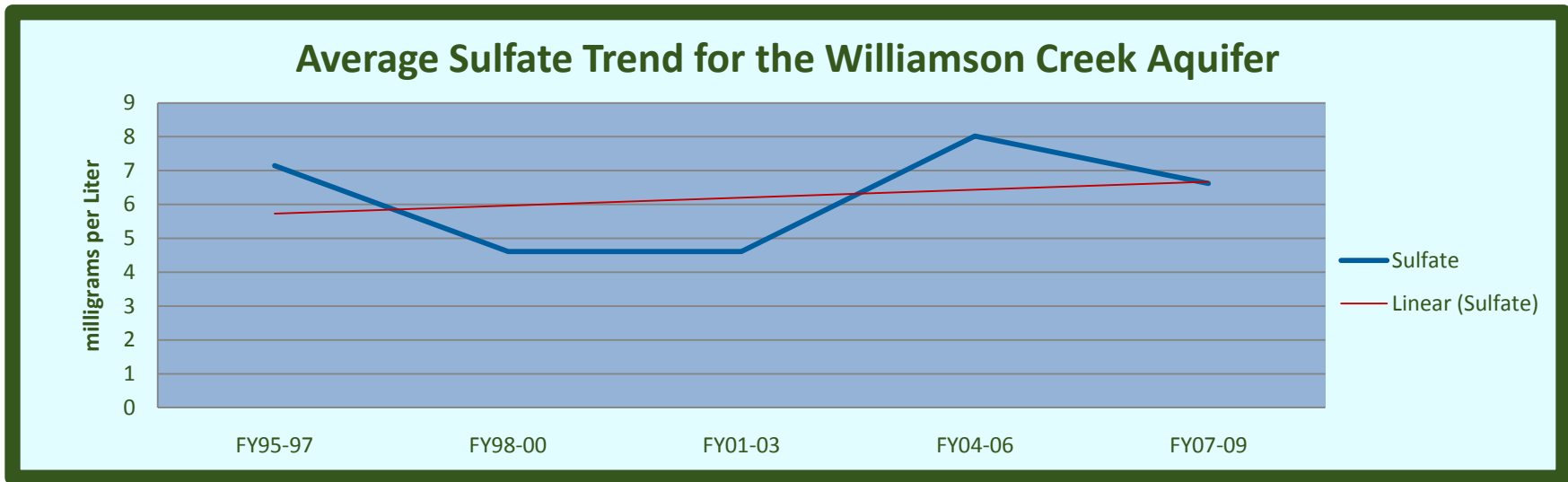


Chart 11-10: Total Dissolved Solids (TDS) Trend

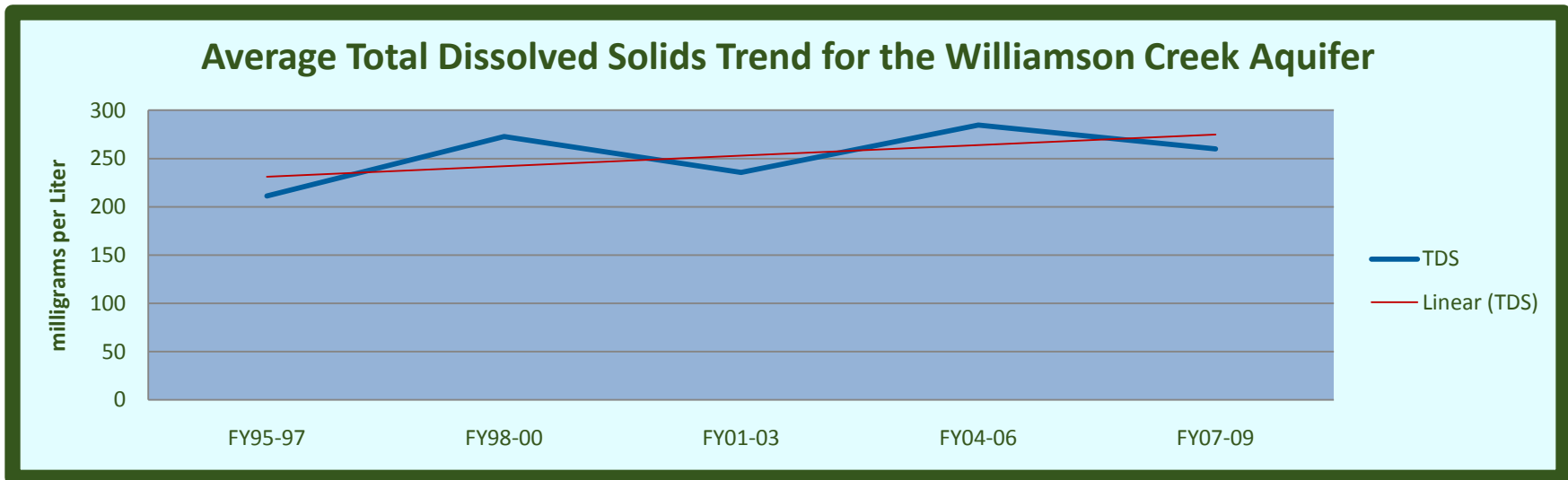


Chart 11-11: Ammonia (NH3) Trend

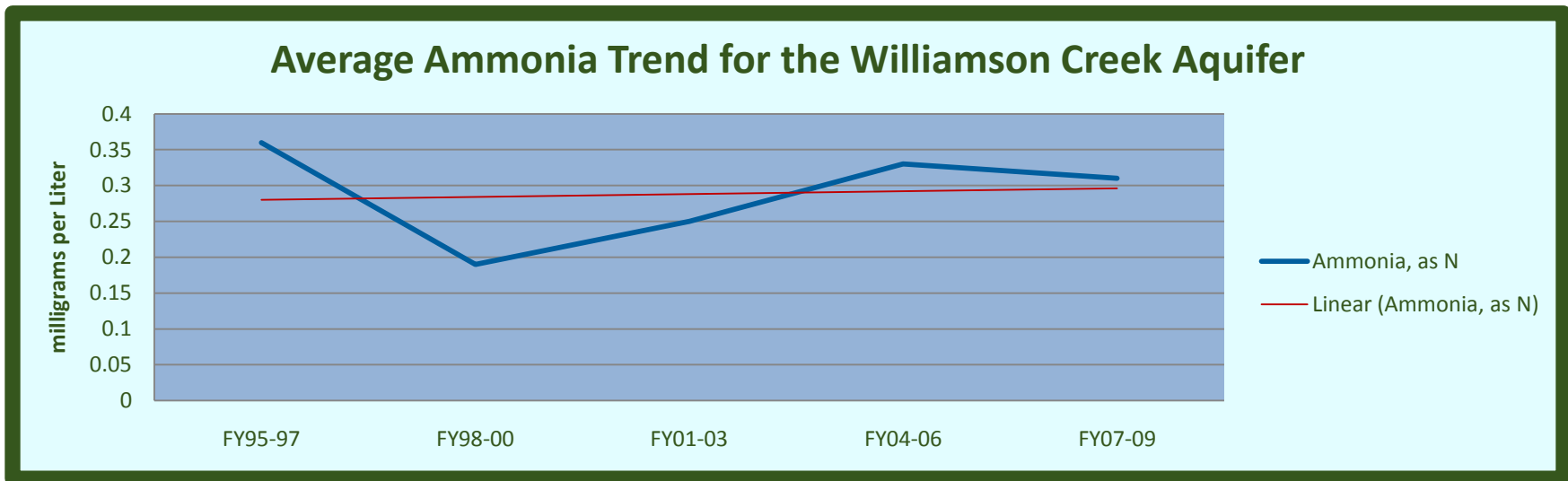


Chart 11-12: Hardness Trend

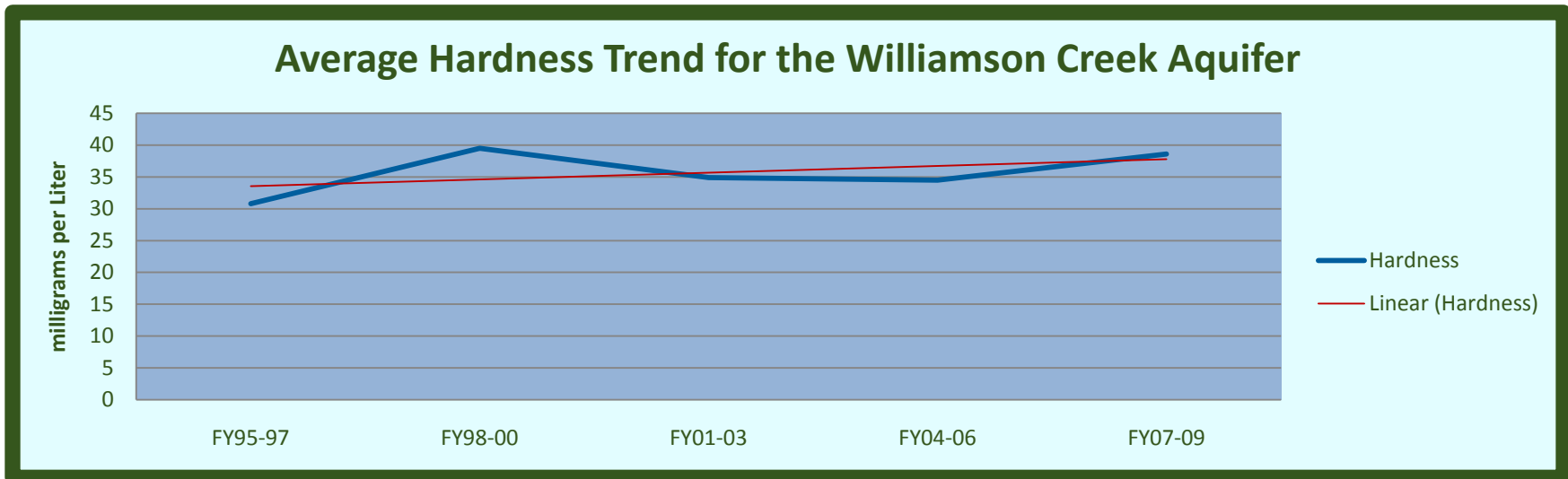


Chart 11-13: Nitrite – Nitrate Trend

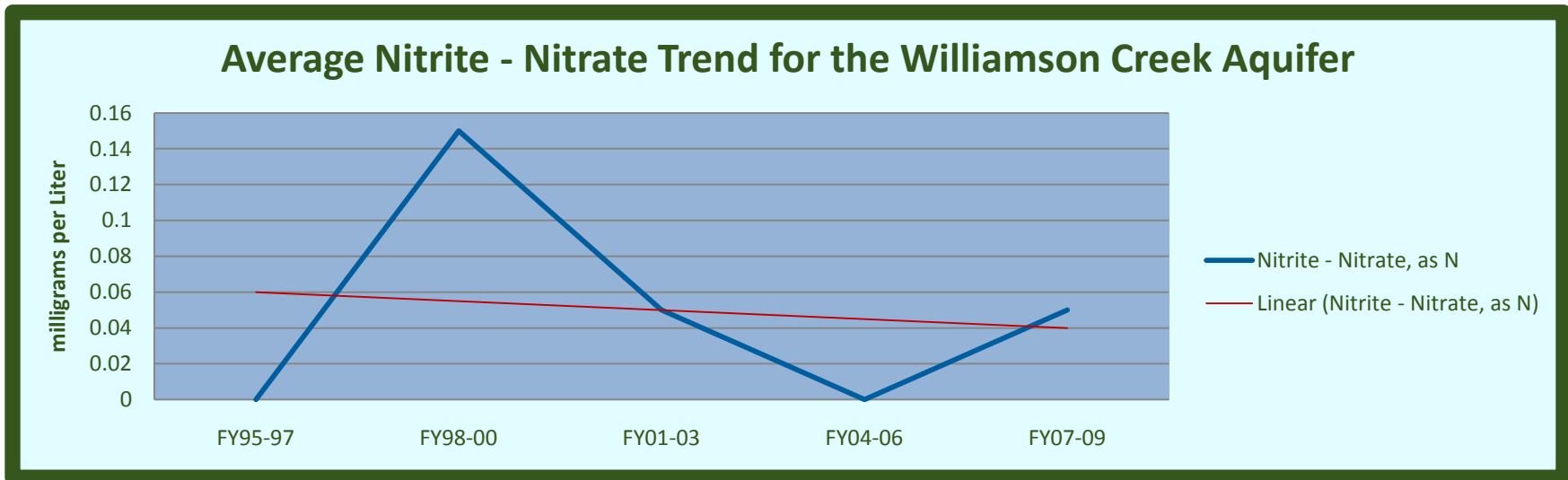


Chart 11-14: TKN Trend

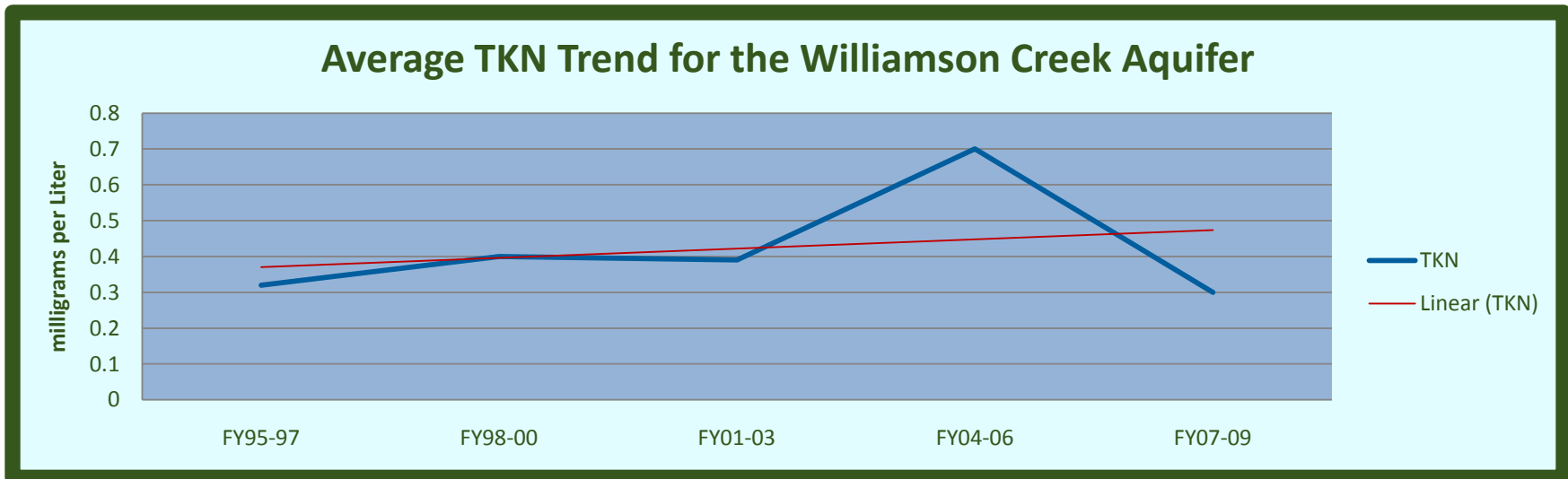


Chart 11-15: Total Phosphorus Trend

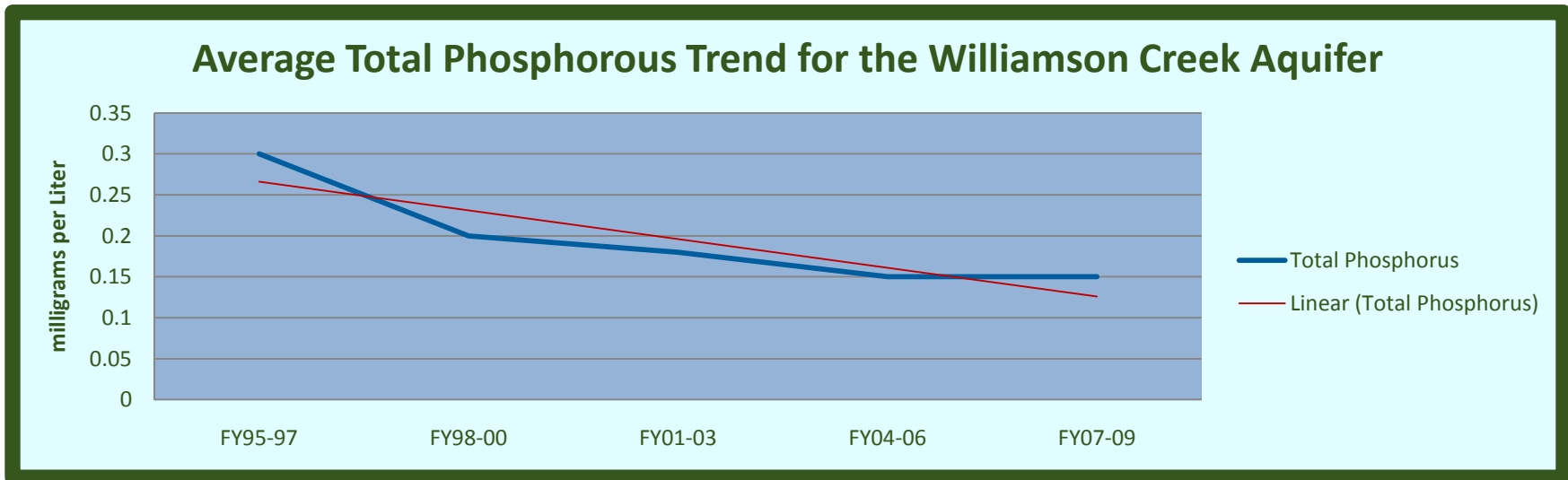


Chart 11-16: Iron Trend

